

## An Efficient Solid-Phase Synthesis of the Vitamin D<sub>3</sub> System

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1 $\alpha$ ,25-Dihydroxyvitamin D<sub>3</sub> (calcitriol) (**1**) is well-known as the hormonally active form of vitamin D<sub>3</sub>, whose physiological activities include regulation of cell differentiation and proliferation, intestinal calcium absorption, bone mobilization, and bone formation.<sup>1</sup> These properties have stimulated significant efforts toward the synthesis of various calcitriol analogues having modified side chains and having some A-ring derivatives.<sup>2</sup> However, there is no report of a convenient method for the preparation of a library of vitamin D<sub>3</sub> analogues having modified side chains and modified A-ring moieties at the same time.<sup>3</sup> We anticipated that a key component for constructing a library of vitamin D<sub>3</sub> analogues is the ability to simply and efficiently couple A-ring moieties, CD-rings, and side chains. The combination of a variety of these three components could provide a large array of analogues if this can be performed on solid-phase resins. Recently, solid-phase synthesis has become a powerful tool for the preparation of not only oligopeptides and oligonucleotides but also small molecule libraries.<sup>4</sup> We report here the novel solid-phase synthesis of the vitamin D<sub>3</sub> system, obtained by efficient coupling of the solid-supported CD-ring with modified A-ring moieties, followed by immediate alkylation with a side chain.

The 11-hydroxy CD-ring **8** is a key intermediate, and our strategy is outlined in Scheme 1. The hydroxy group at the 11-position is used for loading to the polymer resin and the attached group should not affect the subsequent two steps, that is:<sup>5</sup> (i) Horner–Wittig reaction of A-ring moieties **5** to the polymer-supported 8-keto CD-ring **4**,<sup>6–8</sup> (ii) alkylation of the polymer-supported tosylate at the 22-position with the Grignard reagent **6**. The trialkylsilane linker was chosen to attach the polymer support because silyl protection has proven to be effective in the synthesis of various vitamin D<sub>3</sub> analogues and is readily cleaved smoothly even in the presence of the unstable triene moiety. To avoid facile epimerization at the 14-position adjacent to the

carbonyl group, Horner–Wittig reaction of an A-ring moiety to the polymer-supported 8-keto CD-ring is performed initially, giving rise to the triene system of the vitamin D<sub>3</sub> in the presence of a tosylate. Alkylation of a tosyl group at the 22-position of the polymer-support with a Grignard reagent will afford the vitamin D<sub>3</sub> system,<sup>9</sup> if it can be performed at sufficiently low temperature to avoid isomerization of the triene system. ED-71 (**2**) which has three hydroxy units on the A-ring is a promising candidate for osteoporosis therapy<sup>10</sup> and has regulatory activities for calcium metabolism. Contemplating the construction of a big library of A-ring analogues by changing either the alkoxy group or the stereogenic centers, we attempted solid-phase syntheses of 11-hydroxylated ED-71 analogue **3a** and 1 $\alpha$ ,25-(OH)<sub>2</sub>-vitamin D<sub>3</sub> analogue **3b**.

The A-ring moiety **5a** was prepared as follows. Nitrile **9** was synthesized according to our previously reported procedure.<sup>11</sup> Protection of the hydroxy group of **9** by TBS, followed by DIBAL reduction of the nitrile afforded an aldehyde, which was oxidized to acid **10**. Acid **10** was treated with carbonyl diimidazole, followed by the addition of magnesium ethyl malonate, to provide  $\beta$ -keto ester **11**.<sup>12</sup> The (*Z*)-enolate<sup>13</sup> formed by the treatment of **11** with sodium hydride was trapped as a triflate, and the ethyl ester was reduced with DIBAL to yield alcohol **12**. Palladium(0)-catalyzed cyclization was carried out as previously reported.<sup>14</sup> The reaction proceeded smoothly at room temperature to furnish cyclized (*Z*)-diene **13** in 87% yield, exclusively, without formation of (*E*)-diene.<sup>13</sup> Dienyl alcohol **13** was chlorinated (MsCl/LiCl/DMF), and was converted to the phosphine oxide **5a** according to the literature procedure<sup>6–8,15</sup> (Scheme 2).

The  $\beta$ -hydroxyketone **8** was prepared from the Inhoffen–Lythgoe diol,<sup>16,17</sup> and was loaded on chlorinated PS-DES resin (0.74 mmol/g).<sup>18</sup> The loading yield was determined to be 66% by cleavage with HF·Py in THF from the solid support.<sup>19</sup> Horner–Wittig reaction of resin **4** with lithiated **5a** in THF at –78 °C to –40 °C yielded triene **14**. Acid cleavage gave the corresponding 11-hydroxy compound, whose characterization showed exclusive formation of the triene system and no epimerization at the 14-position. Coupling of tosylate **14** with Grignard reagent **6** at –10

(9) In most of precedent syntheses of vitamin D<sub>3</sub> derivatives, Grignard coupling reaction was initially carried out with a CD-ring that possesses not 8-keto but 8-hydroxy group. Then, the hydroxy group was oxidized to provide an 8-keto compound, which underwent Horner–Wittig olefination leading to the vitamin D<sub>3</sub> system. There are only a few examples for alkylation of tosylate at the 22-position in the presence of triene system with Grignard reagent. Especially, few examples of alkylation have been reported using 3-alkoxy-3-methylbutyl Grignard reagent for the direct formation of the side chain of vitamin D<sub>3</sub> system; see: Andrews, D. R.; Barton, D. H. R.; Hesse, R. H.; Pechet, M. M. *J. Org. Chem.* **1986**, *51*, 4819.

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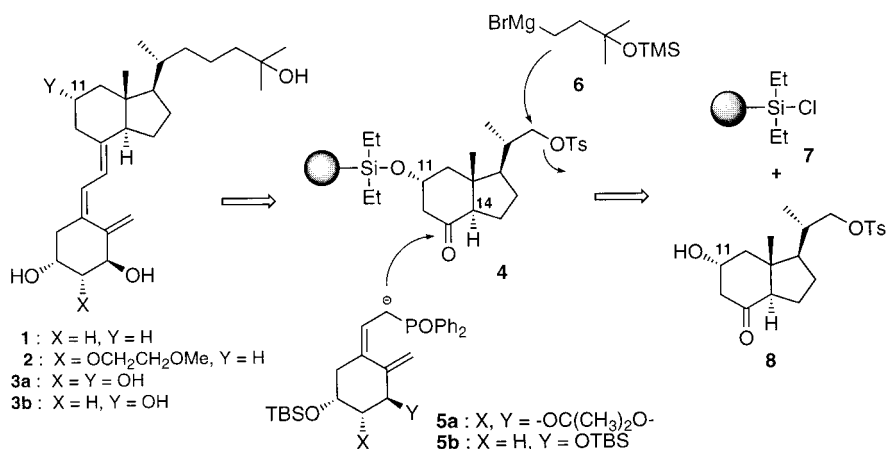
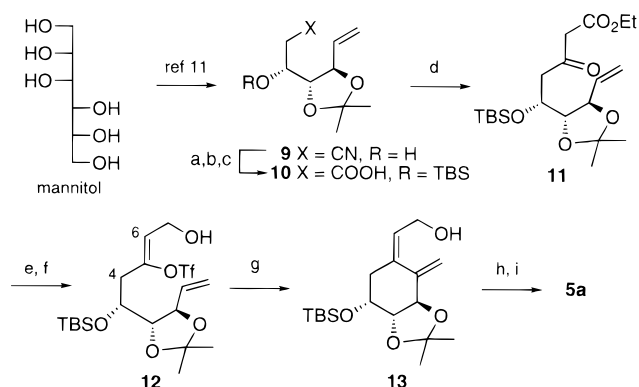
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## Scheme 1

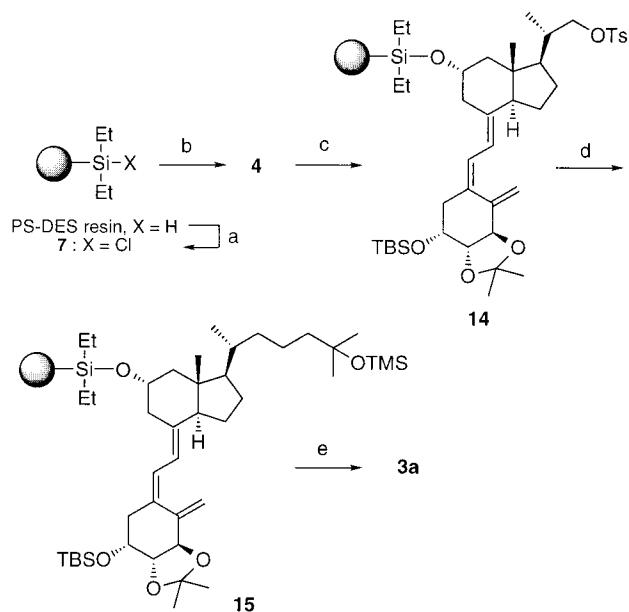
Scheme 2<sup>a</sup>

<sup>a</sup> (a) TBSCl, imidazole, 98%; (b) DIBAL, 83%; (c) NaClO<sub>4</sub>, NaH<sub>2</sub>PO<sub>4</sub>, 2-methyl-2-butene, 100%; (d) CDI, (EtOCOCH<sub>2</sub>CO<sub>2</sub>)<sub>2</sub>Mg, 99%; (e) NaH, Tf<sub>2</sub>NPh, THF, 63%; (f) DIBAL, 96%; (g) Pd(OAc)<sub>2</sub> (10 mol %), PPh<sub>3</sub> (20 mol %), NEt<sub>3</sub> (2 equiv), DMF, rt, 87%; (h) MsCl, LiCl, lutidine, 85%; (i) *n*-BuLi, Ph<sub>2</sub>PH, THF; 5% H<sub>2</sub>O<sub>2</sub>, 81%.

°C led to the alkylation product **15**.<sup>20</sup> Finally, treatment of **15** with HF·Py in THF at room temperature for 24 h released the desired vitamin D<sub>3</sub> system **3a** (Scheme 3). Simple filtration through silica gel provided a white powder **3a** in 62% overall yield from **4**. The purity of the final product, judging by HPLC analysis, was high, and there is no epimerization at the 14-position of ketone **4** and no triene isomerization under these conditions. This procedure could also be applied to an activated vitamin D<sub>3</sub> analogue. Starting from **5b**,<sup>7,8</sup> the solid-phase synthesis described above (Horner–Wittig reaction with **4**, alkylation with Grignard reagent **6**, and the cleavage from the resin) afforded **3b** (>95% pure) in 61% overall yield.

We have demonstrated an efficient and general strategy for the solid-phase synthesis of vitamin D<sub>3</sub> systems. This strategy promises to overcome the required two carbon–carbon bond-forming steps on the solid-phase, resulting in the consecutive coupling of the CD-ring with the A-ring and side-chain moieties.

(20) The alkylation by means of the chloride salt of the Grignard reagent instead of the bromide salt **6** was unsuccessful. There are few reports using an alkoxy Grignard reagent in the alkylation at the 22-position on the steroid system, see: Nakagawa, M. Ph.D. thesis, University of Tokyo, 1993 and ref 9.

Scheme 3<sup>a</sup>

<sup>a</sup> (a) 1,3-Dichloro-5,5-dimethylhydantoin (3 equiv), CH<sub>2</sub>Cl<sub>2</sub>, rt, 1 h; (b) **8** (4 equiv), imidazole (4.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, rt, 6 h (66%); (c) **5a** (8 equiv), *n*-BuLi (7.5 equiv), THF –78 to –40 °C, 4 h; (d) **6** (30 equiv, 0.45 M), CuBr·Me<sub>2</sub>S (3 equiv), THF, –10 °C, 6 h; (e) HF·Py, THF, rt, 24 h, 62% (from resin **4**).

This solid-phase synthesis will be a powerful method for the preparation of a variety of vitamin D<sub>3</sub> derivatives.

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**Supporting Information Available:** Experimental information, including experimental procedures for **5a**, **3a**, and **3b**, 270 MHz <sup>1</sup>H NMR and 67.8 MHz <sup>13</sup>C NMR spectra of **3a**, **3b**, **5a**, **9–13**, IR spectra of resins **4**, **14**, and **15**, and details of the preparation of **8** (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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